Factors impacting the process and system performance Or How to help the methanotrophs do their job

Julia Gebert

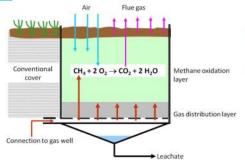
Department Geoscience and Geo-Engineering Delft University of Technology

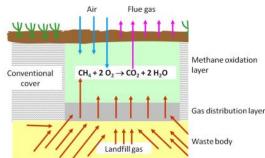


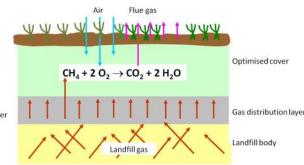


CH₄ oxidation systems: Design goals









Design goals

- Methylobacter sp.
 0.5 µm

 biomass

 CH₄ + 2 O₂

 CO₂ + 2 H₂O + energy
- (1) Adequate physicochemical environment of high structural stability
- (2) Optimal diffusive ingress of oxygen
- (3) Maximum spatial evenness of gas load
- (4) Robust dimensioning of the system, adapted to load

Choice also depends on

- Intention of measure (e.g. safety, climate)
- After-use of landfill (e.g. access for the public?)

Physical properties of material



(1) Create adequate physicochemical environment of high structural stability

Aims:

- Support biological activity for both bacteria and vegetation
- Avoid loss of permeability and formation of preferential pathways



Requirements properties of MOL

Parameter	Value	Meaning
Soil pH	5.5 to 8.5	Optimum MOB
El. conductivity	< 4 mS/cm	Avoid osmotic stress
Plant-available water	14 vol.%	Support vegetation and MOB
Air-filled porosity at field capacity	14 vol.%	Diffusion of O ₂
Organic matter	2 to 4%, 8% if stable	Nutrient supply to MOB and vegetation
Low susceptibility to consolidation	Preservation of pore structure	
Low susceptibility to cracking	Avoid preferential pathways	



Conclusions adequate geophysical environment

- Conditions for methanotrophic bacteria can be met by a wide range of materials
- What is good for the vegetation, is good for methanotrophic bacteria (nutrients, water)
- Special attention for aeration
- No nitrogen fertilizer or nitrogen-rich amendments due to inhibitory effects of NH₄⁺
- Organic materials (keynote 2) have to be stable
 - \rightarrow minimize competition for O₂
 - → minimize settlement and loss of permeability





(2) Optimize diffusive ingress of oxygen

Aim:

Maximize depth of aeration to

- Create thick and "redundant" CH₄-oxidation layer
- render oxidation process less susceptible to surface effects (frost, drought, heat, cold)

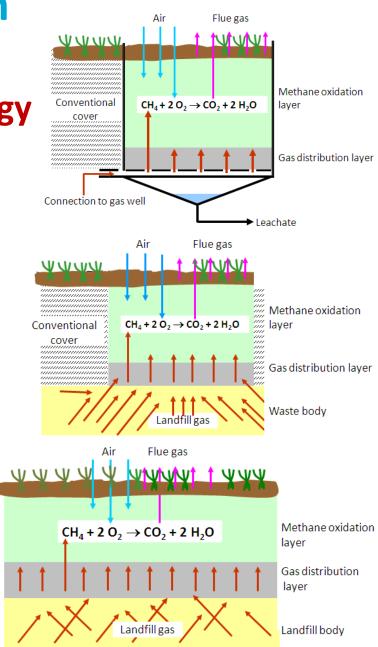


Optimize ingress of oxygen

$$CH_4 + 2O_2 \rightarrow CO_2 + 2H_2O + energy$$

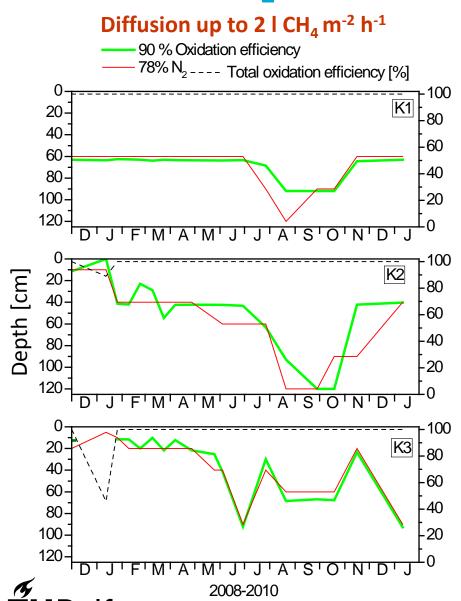
- Twice the volume of O₂ is needed for complete oxidation
- O₂ is provided only from the atmosphere
- Main driver is the concentration gradient, main transport process is diffusion
- → Effective diffusivity of the soil is absolutely crucial



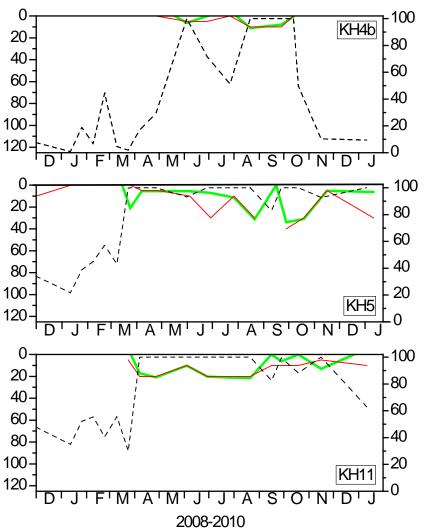


Oxidation efficiency[%]

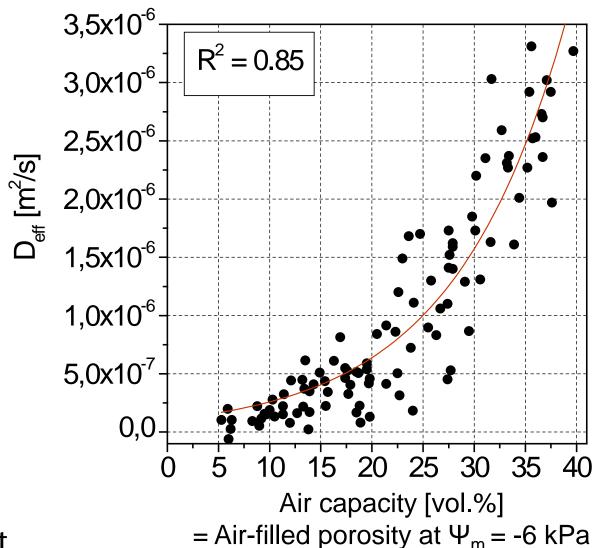
Relevance of O₂ supply



Advection up to 60 l CH₄ m⁻² h⁻¹

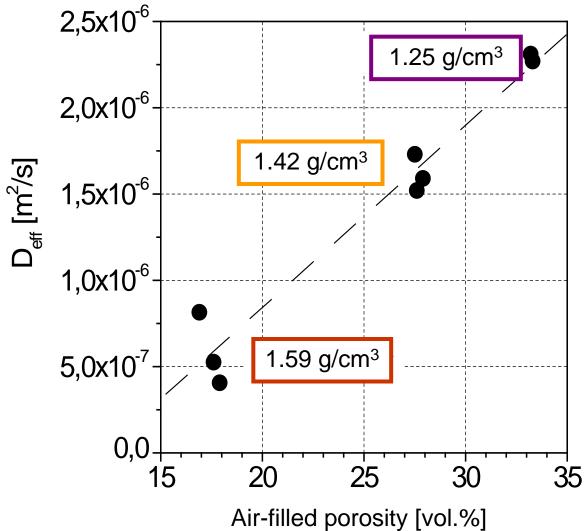


Diffusivity depends on air-filled porosity





Compaction decreases diffusivity



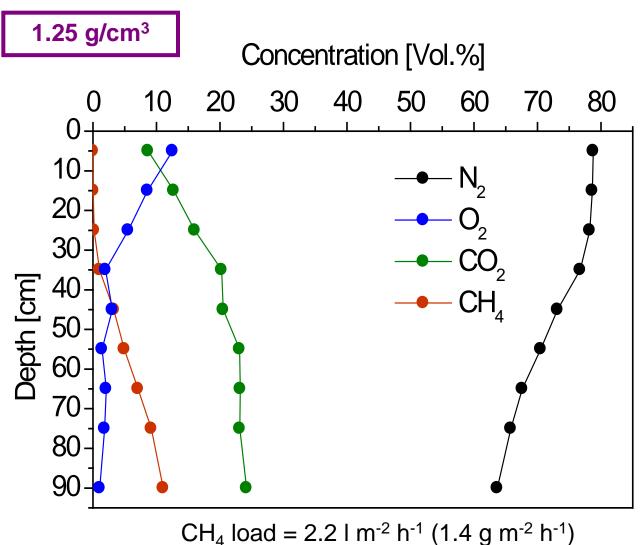


Column experiment



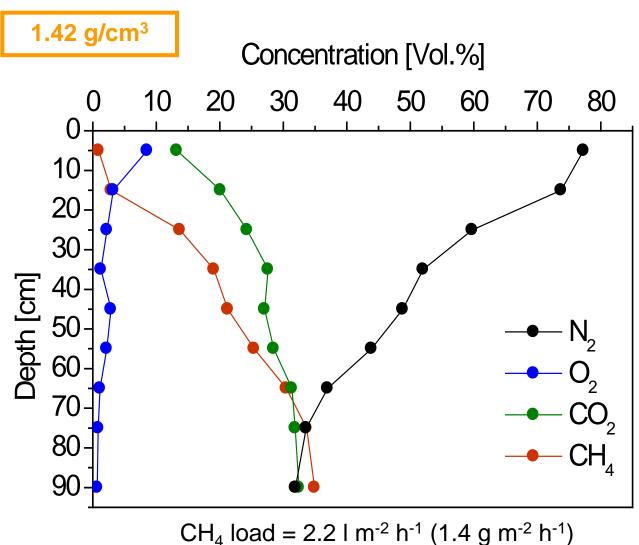


Gas profiles with compaction \(\)



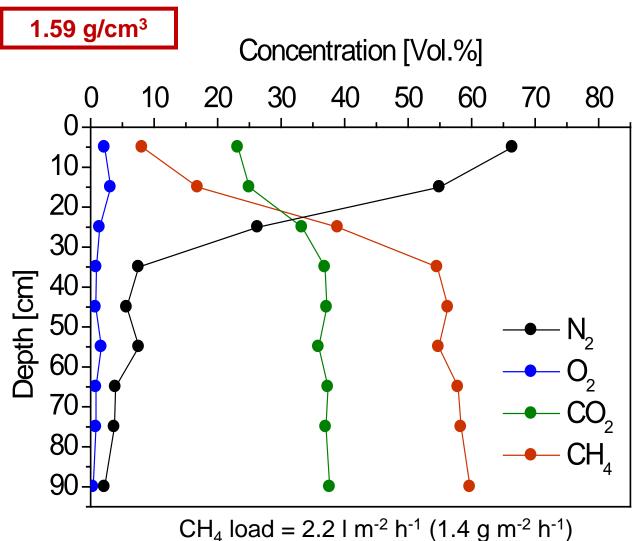


Gas profiles with compaction ↑



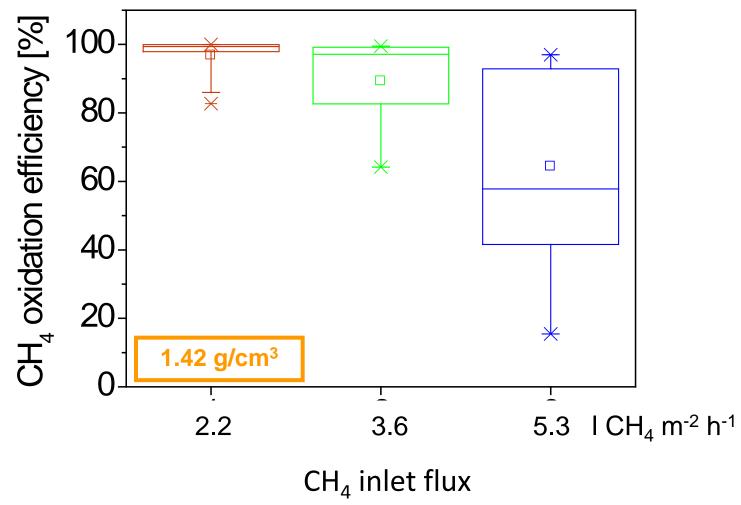


Gas profiles with compaction ↑



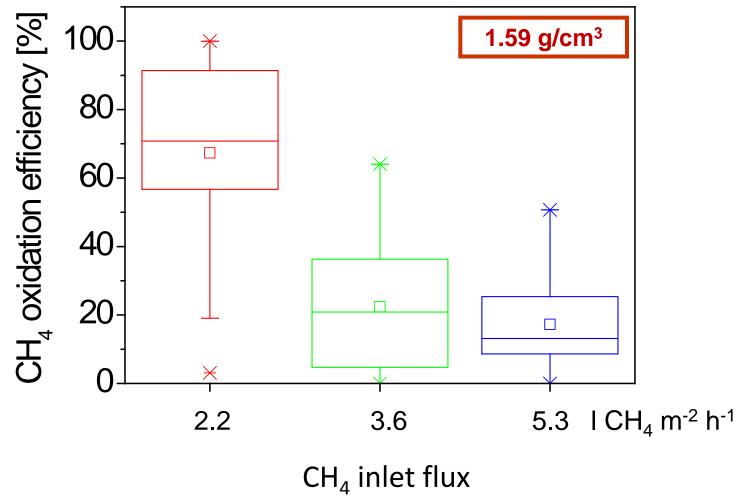


Oxidation efficiency with advection \(\), medium bulk density



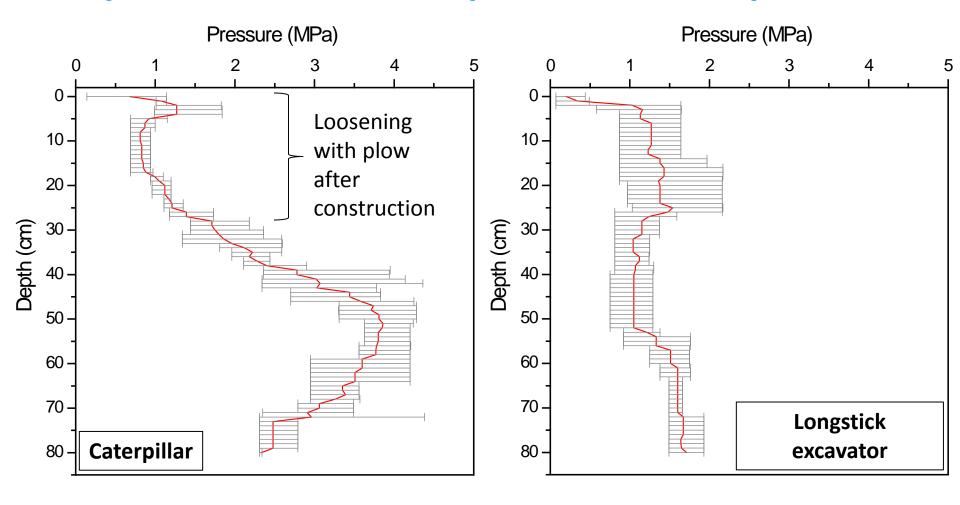


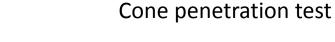
Oxidation efficiency with advection \(\), high bulk density





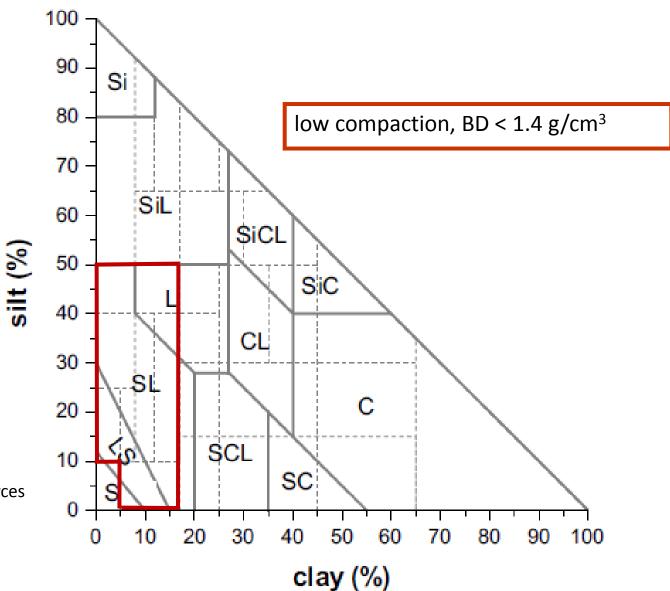
Impact of construction practice on compaction







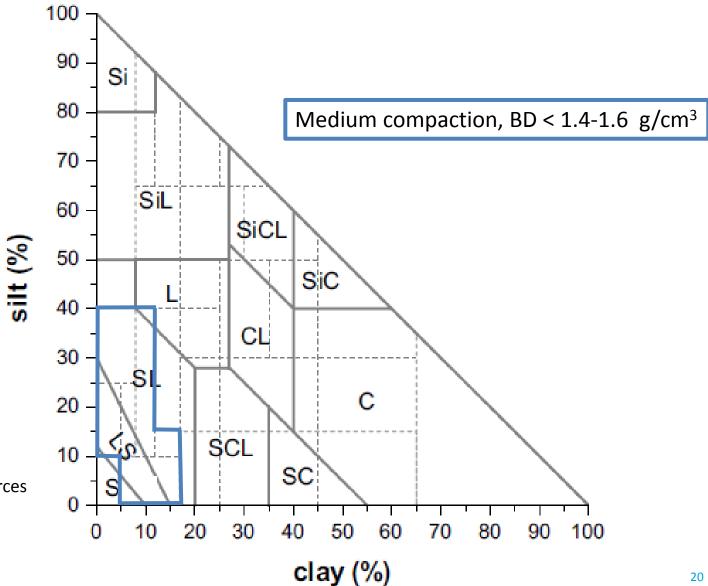
Soil textures meeting target of 14 vol.% AFP



FAO/ISS (2006): World reference base for soil resources



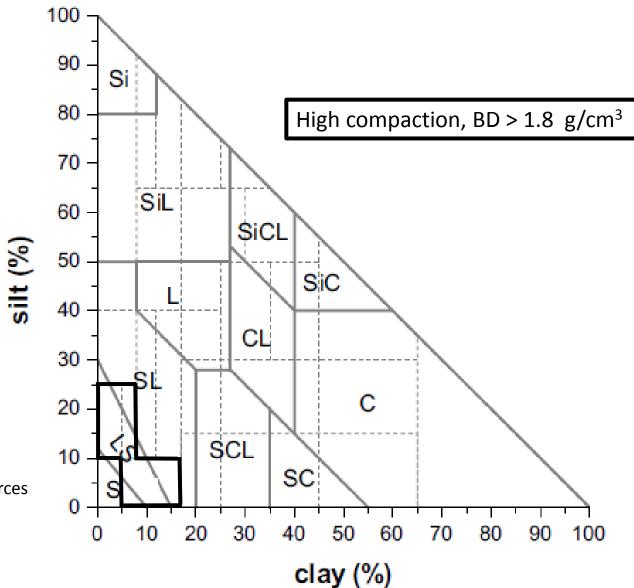
Soil textures meeting target of 14 vol.% AFP



FAO/ISS (2006): World reference base for soil resources



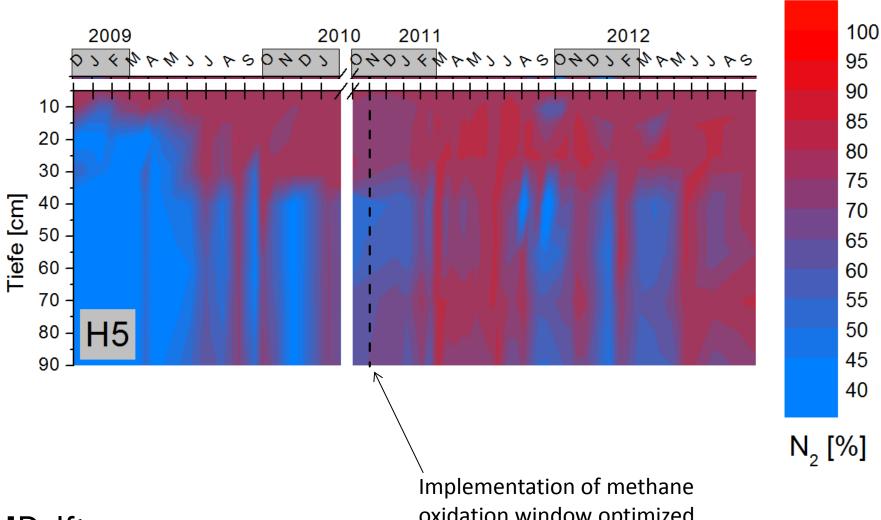
Soil textures meeting target of 14 vol.% AFP



FAO/ISS (2006): World reference base for soil resources



Enhancing aeration by hotspot remediation





oxidation window optimized for aeration

Conclusions O₂ supply

- Process heavily dependent on adequate O₂ supply
- O₂ diffusive ingress governed by soil effective diffusivity
- Diffusivity depends on air-filled porosity at given water content and hence on texture and compaction
- Empirical relationships allow for good estimate
- Choice of suitable soil textures and construction practice is crucial

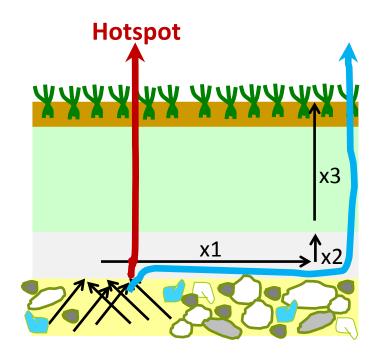








(3) Spatial evenness of gas load

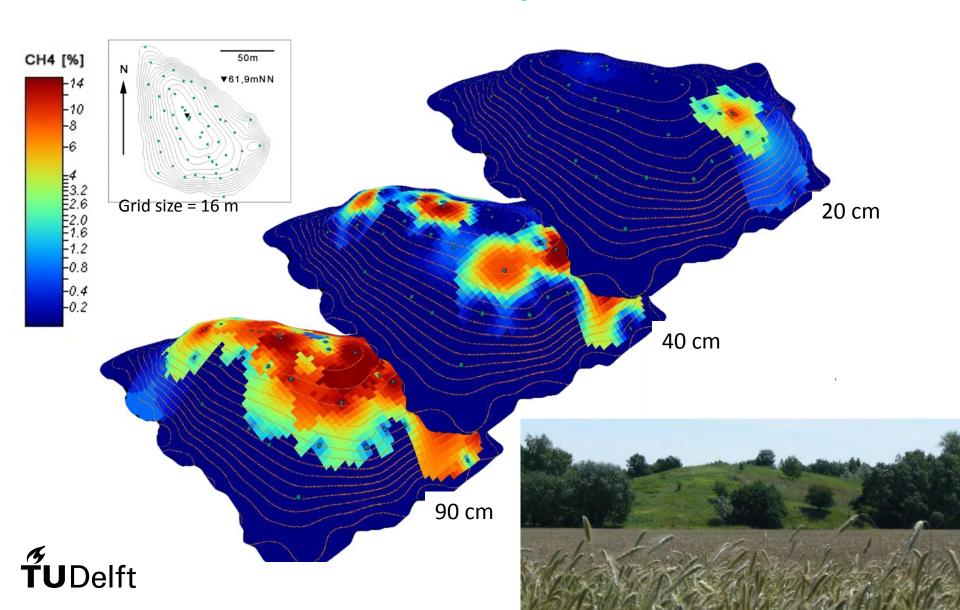


Aims:

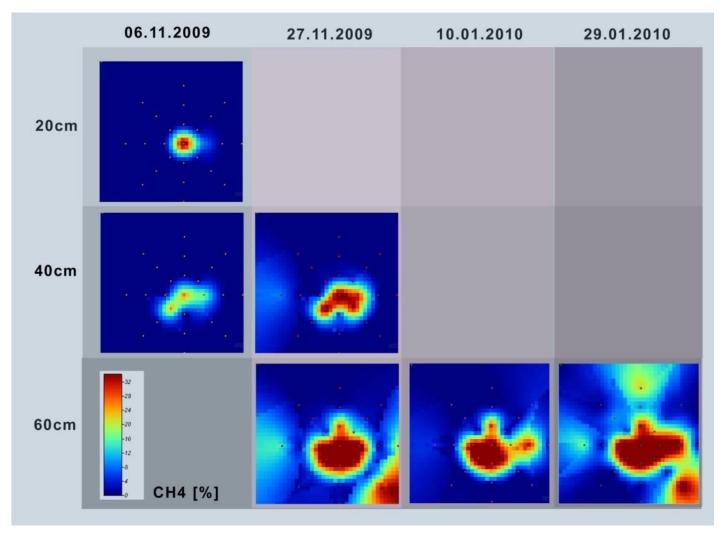
- Avoid overloading of individual compartments
- Tap full system potential
- Avoid channelled advective transport



Spatial variability of soil gas composition in a cover soil of a non-sanitary landfill



Methane concentration at hotspot



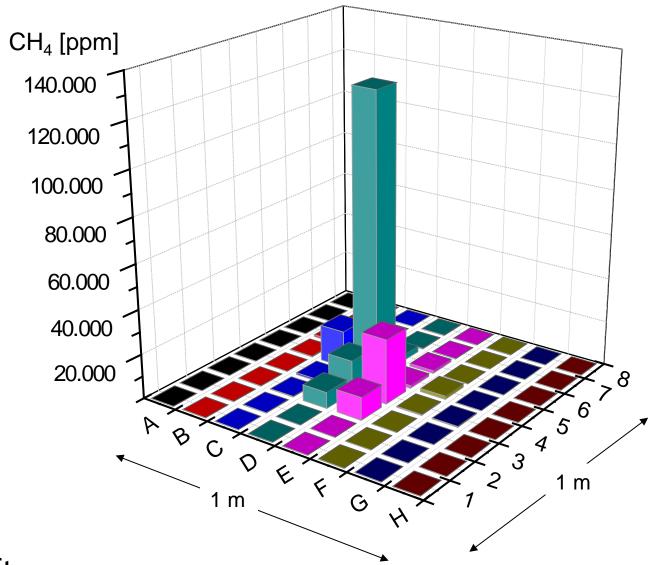


Morphology of hotspot soil profile



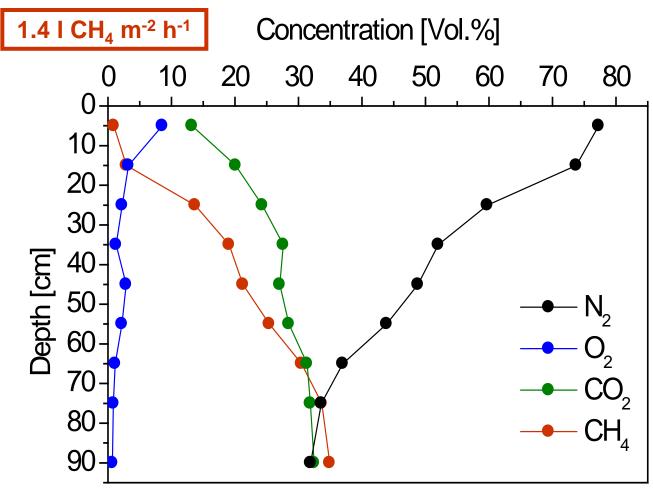


Small scale variability of surface CH₄ concentrations





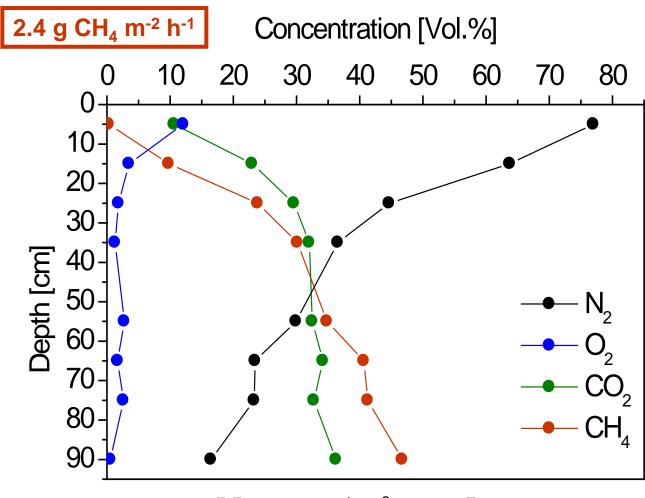
Gas profiles with advection ↑





 $BD = 1.59 \text{ g/cm}^3, 85 \% \text{ Proctor}$

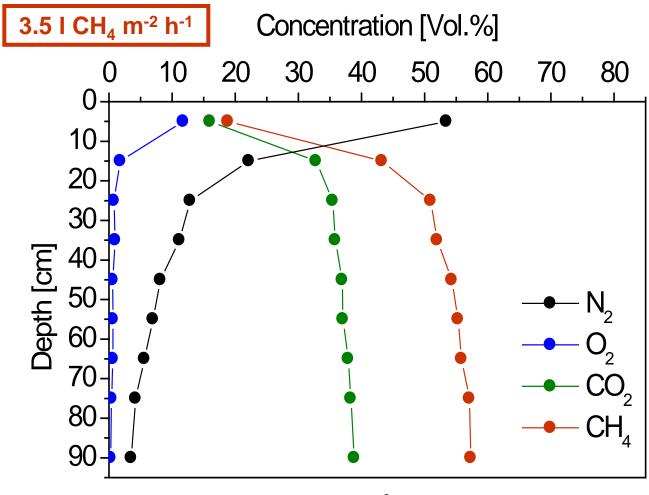
Gas profiles with advection ↑





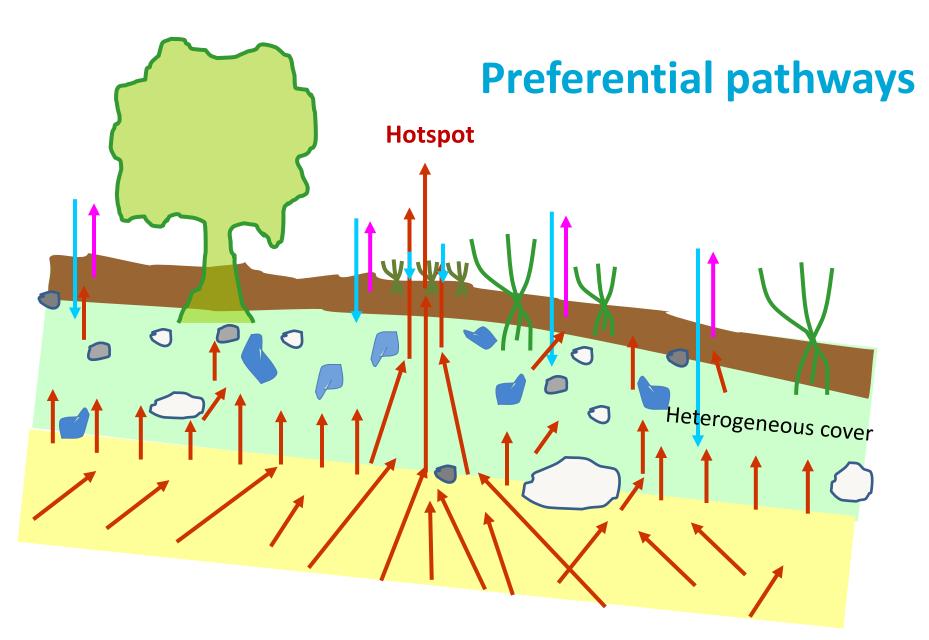
 $BD = 1.59 \text{ g/cm}^3, 85 \% \text{ Proctor}$

Gas profiles with advection \(\)





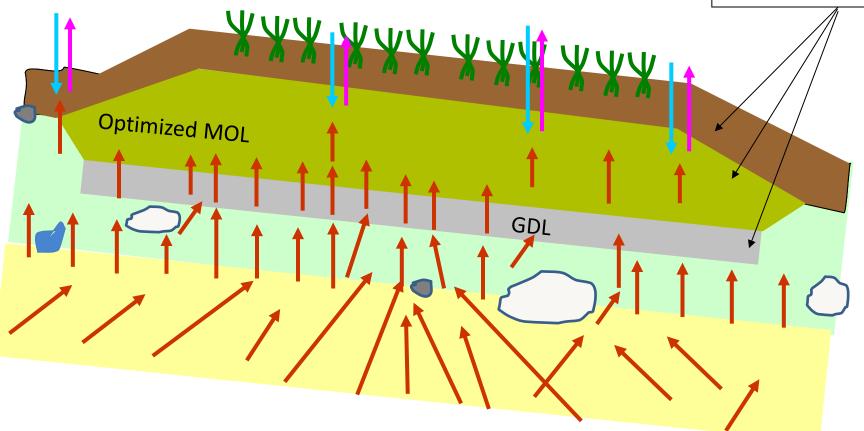
 $BD = 1.59 \text{ g/cm}^3, 85 \% \text{ Proctor}$

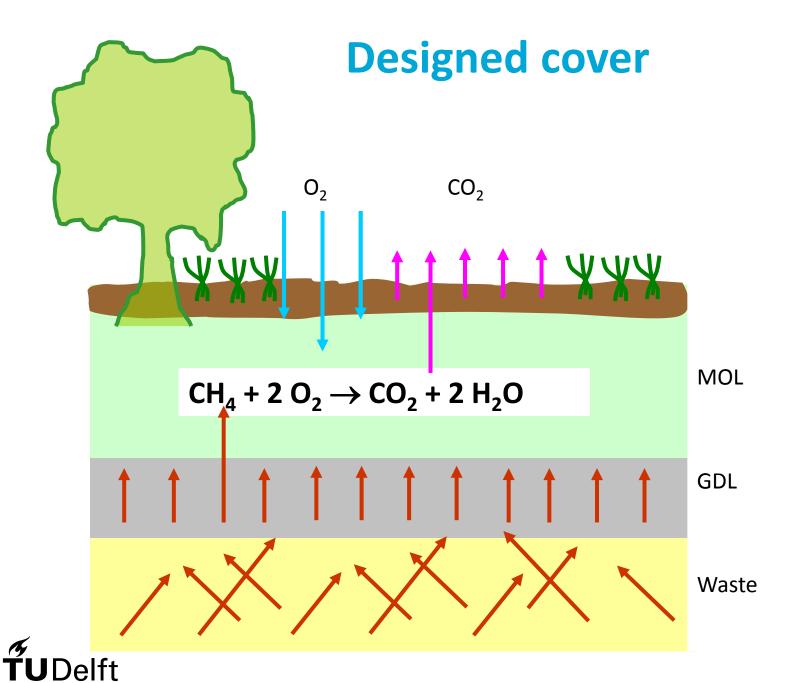




Remediated hotspot

Define properties and construction practice

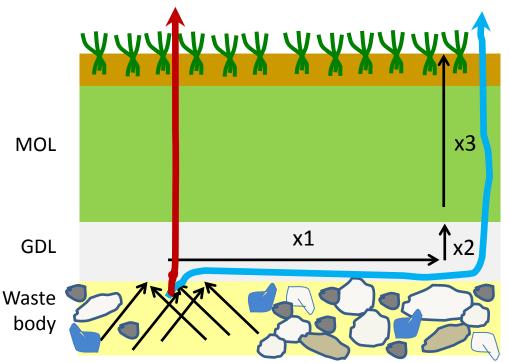




Requirements gas distribution layer

- 1. $< 2\% CaCO_3$
- 2. Purely mineral
- 3. High gas coductivity

- → Avoid precipitation of CO₂
- → High structural stability
- \rightarrow $k_{Gas\ GDL}$ >> $k_{Gas\ MOL}$, so that



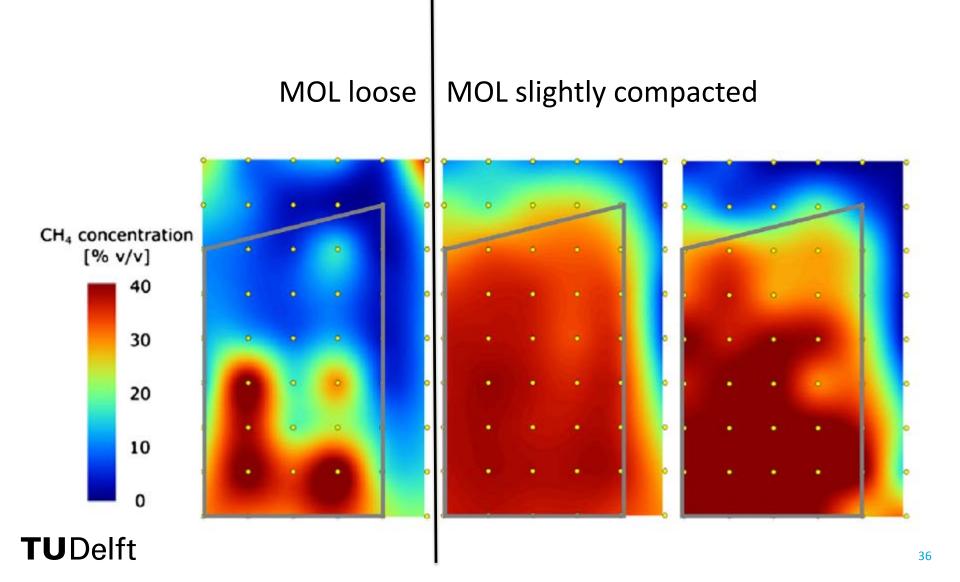
$\Sigma(R_{x1+x2+x3}) >> \Sigma(R_{x1})$

- → Sum resistance should be homogenous over all path lengths
- → horizontal gas transport favoured in GDL

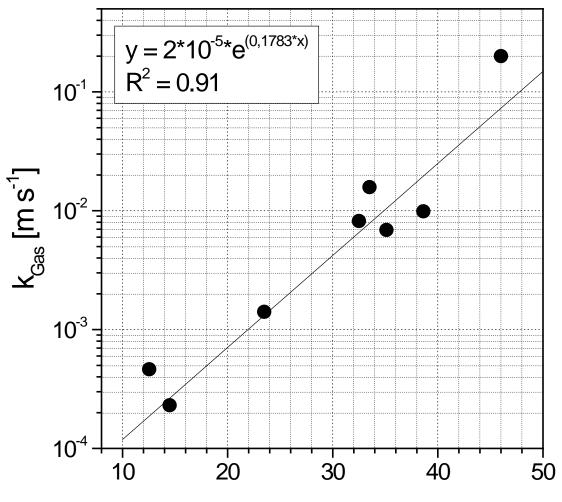
With $R = 1/k_{Gas}$



Impact of decreasing kgas in the MOL



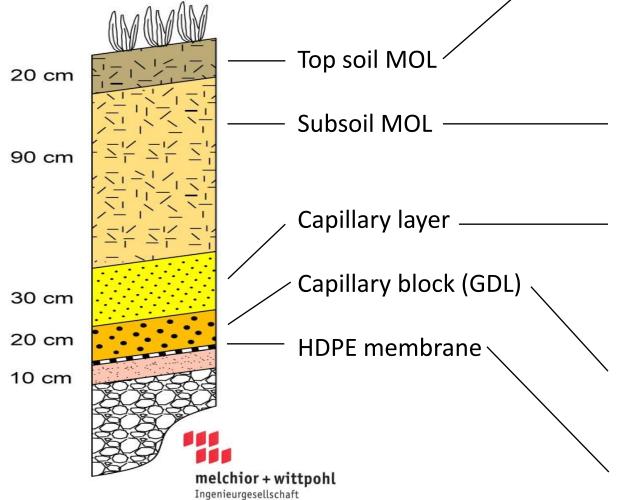
Conductivity (advection) depends on air-filled porosity





Air capacity [vol.-%] = Air-filled porosity at Ψ_m = -6 kPa

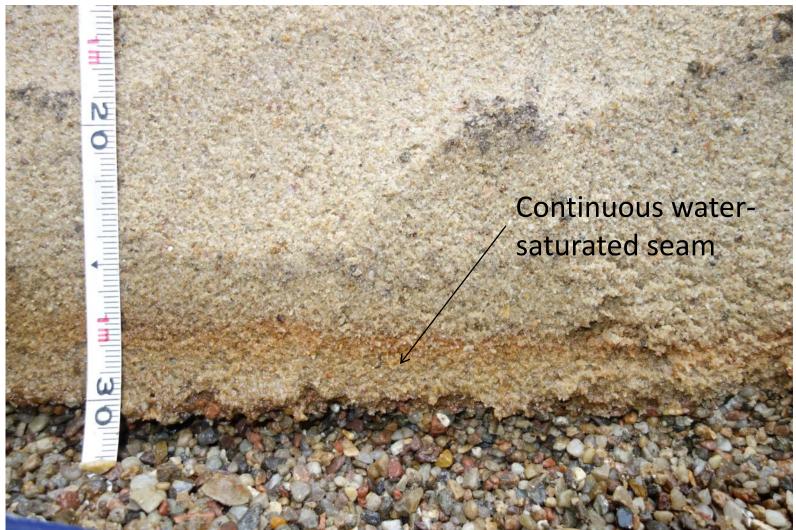
Gas distribution on a slope







Detail capillary layer



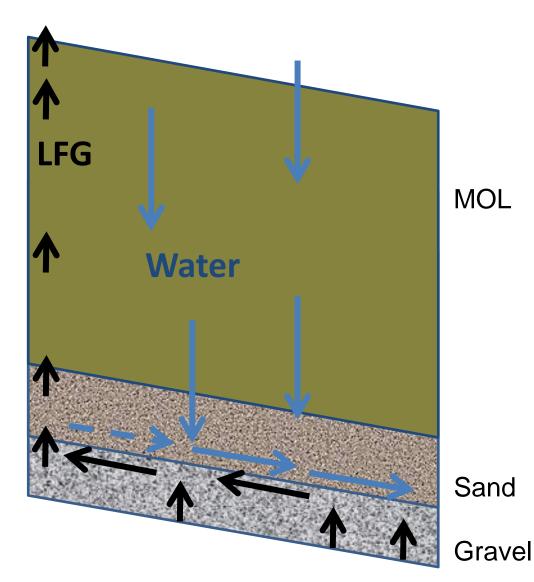


Combination of CH₄ oxidation and water

diversion on slopes

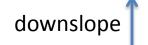
Downslope movement
 of water following
 gravity leads to a
 closed capillary seam
 at foot of slope

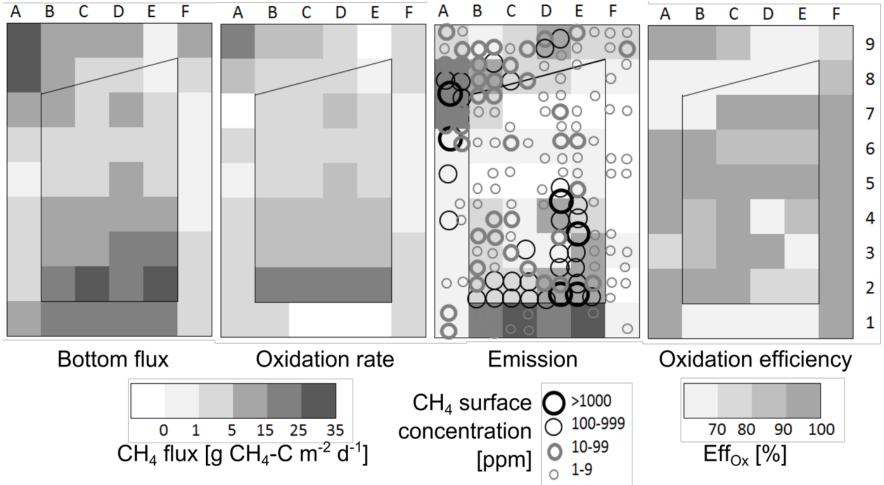
 Gas travels upslope along preferential gas pathway





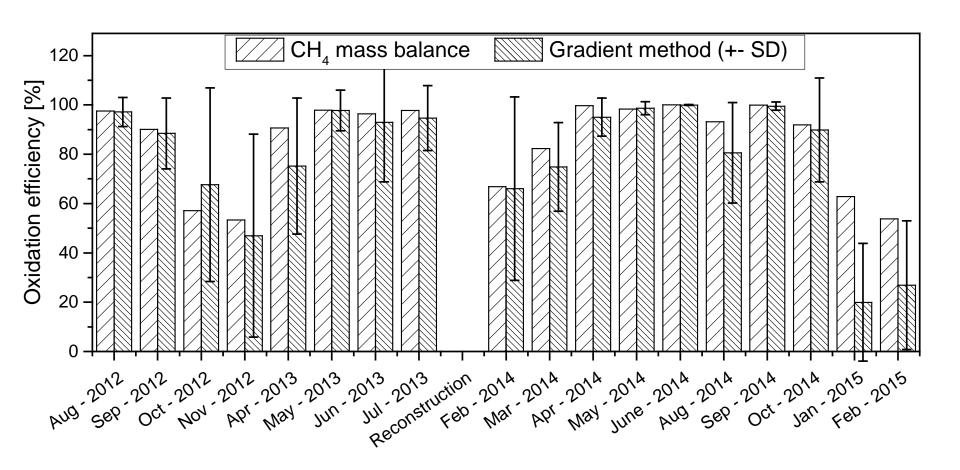
CH₄ fluxes in relation to slope







Ox. eff. of the whole test field still 84%!

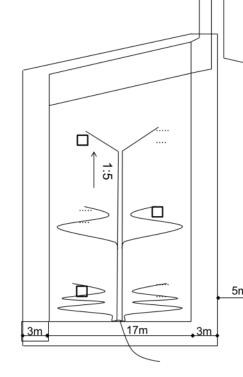


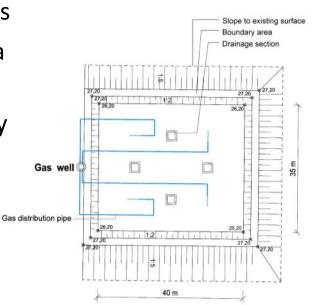


 CH_4 -load: 35 g m-2 d-1

Conclusions spatial distribution

- Gas distribution layers are an essential element of MOS design
- Spatial evenness of gas load depends on difference in gas conductivity between GDL and MOL
- Maximizing this difference is limited by the requirement on diffusivity for the MOL
 - → Calculate pressure losses over path lengths
 - → Decide on maximum difference in pressure loss
 - → Define number of gas inlet points per unit area
- If system is on slope, account for higher necessary oxidation capacity upslope







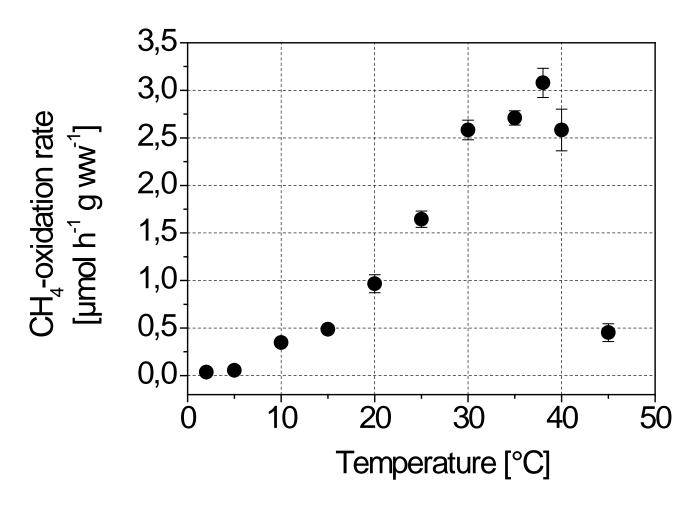
(4) Dimension adapted to load

Aims:

- Decrease spatial load to below the expected spatial CH₄ oxidation potential
- Consider seasonal variation of oxidation rate (temperature and saturation)

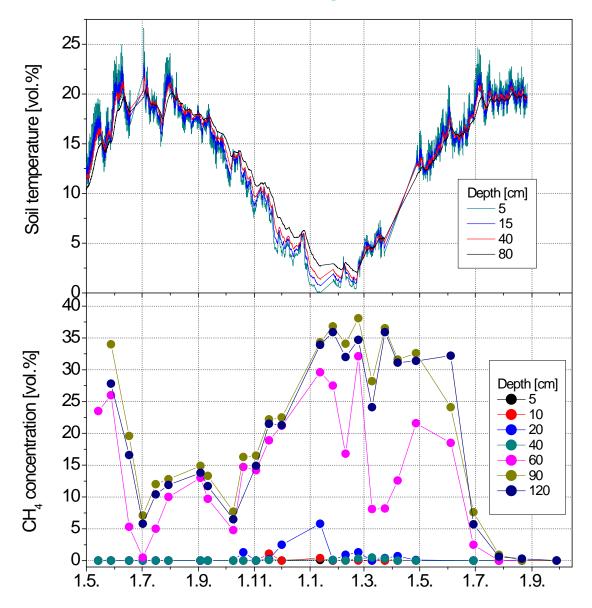


Impact of temperature



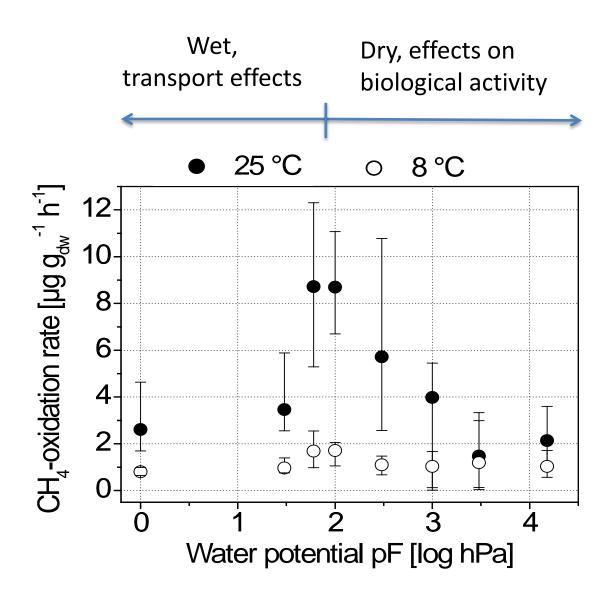


Seasonal effects: Temperature





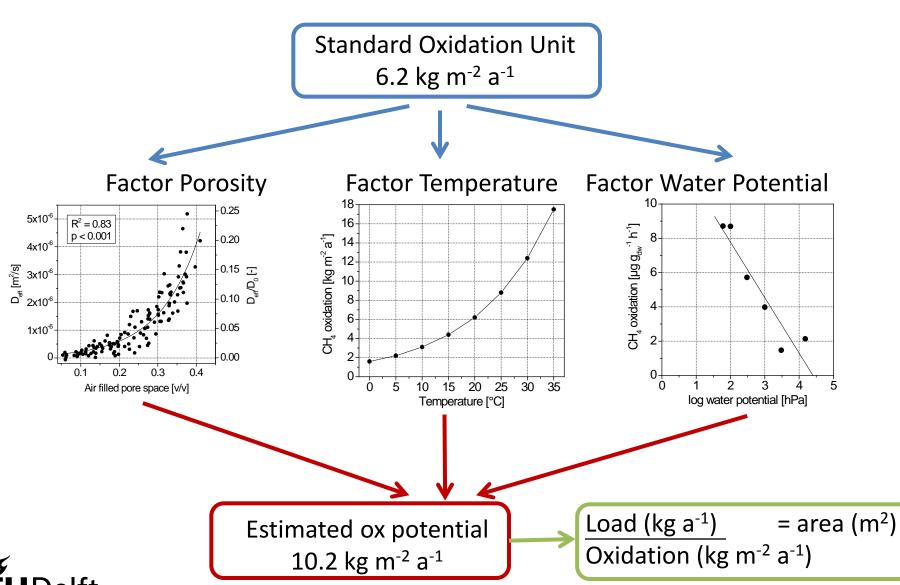
Impact of water potential





Gebert, 2013 47

Methane Oxidation Tool (MOT)





Designing for load

- Estimate CH₄ oxidation potential based on soil properties and climatic conditions
- Design follows limiting factor:
 high quality soil vs. availability of space
- Consider seasonal changes in CH₄ oxidation activity
- Consider required performance
- Given the soil texture, water potential, porosity and hence diffusivity can be easily predicted and modelled



Monitoring



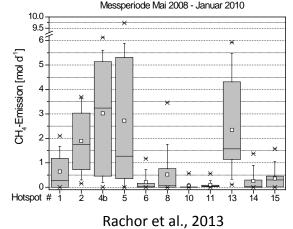
What do you want to know

- Detection of emissive spots?
- Quantification of emissive spots?
- Performance of windows or filters?
- Whole-site emissions?

Point measurements

Spatially integrating measurements

- Consider high spatial and temporal variability of gas fluxes and CH₄ oxidation rates
- Is it research (process information), is it longterm performance monitoring, should it prove on-site safety?
 - different techniques and timely resolution, maybe even limit values









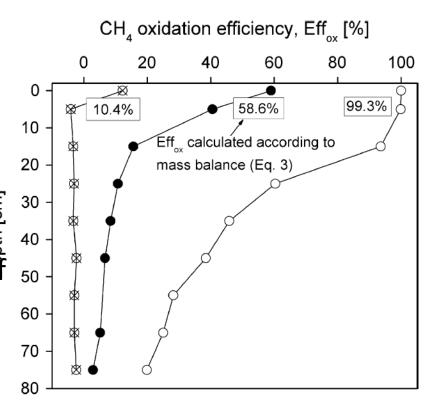
Added value of combined CH₄ and CO₂ measurement

$$CH_4 + 2 O_2 \rightarrow CO_2 + 2 H_2O$$

C-balance: 1 CH₄ goes to 1 CO₂

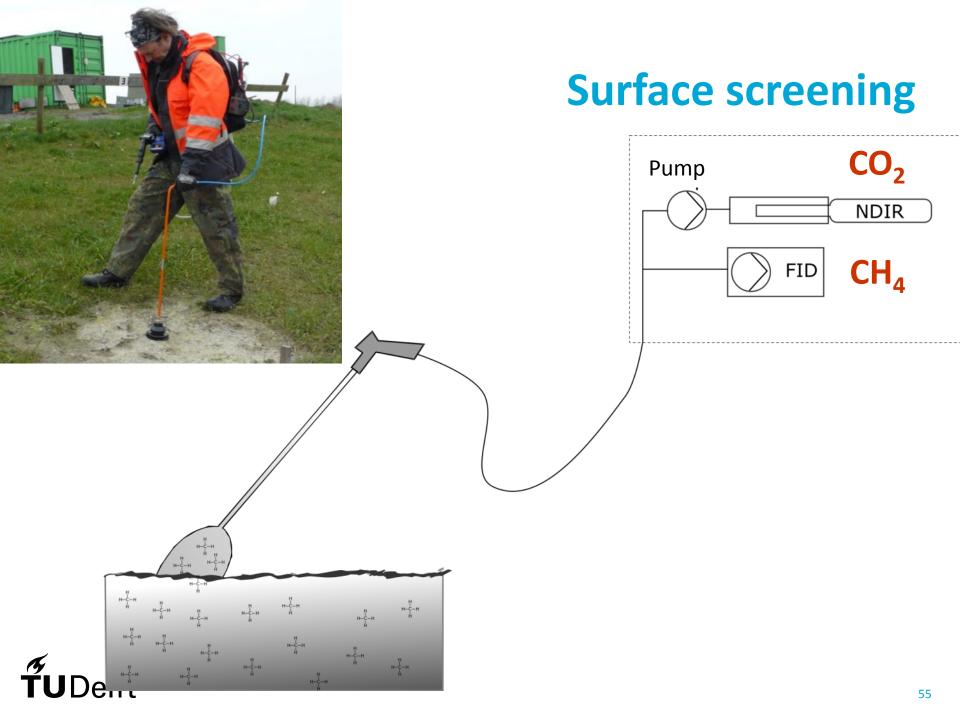
 $CH_4 \downarrow and CO_2 \uparrow$

Ratio $CO_2: CH_4^{2} \uparrow$ Shift of ratio enables calculation of oxidation efficiency (Christophersen et al., 2001)

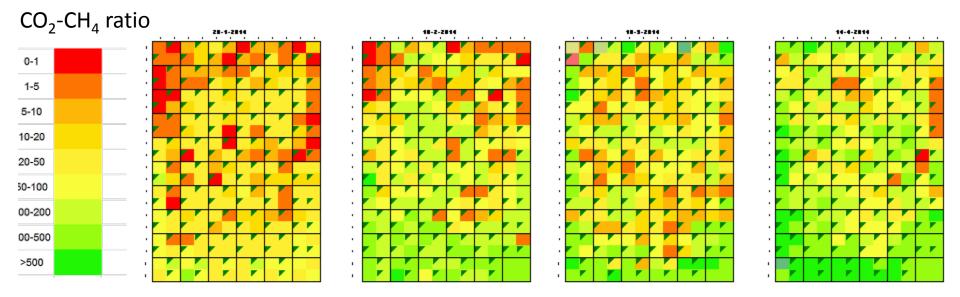


Gebert et al., 2011

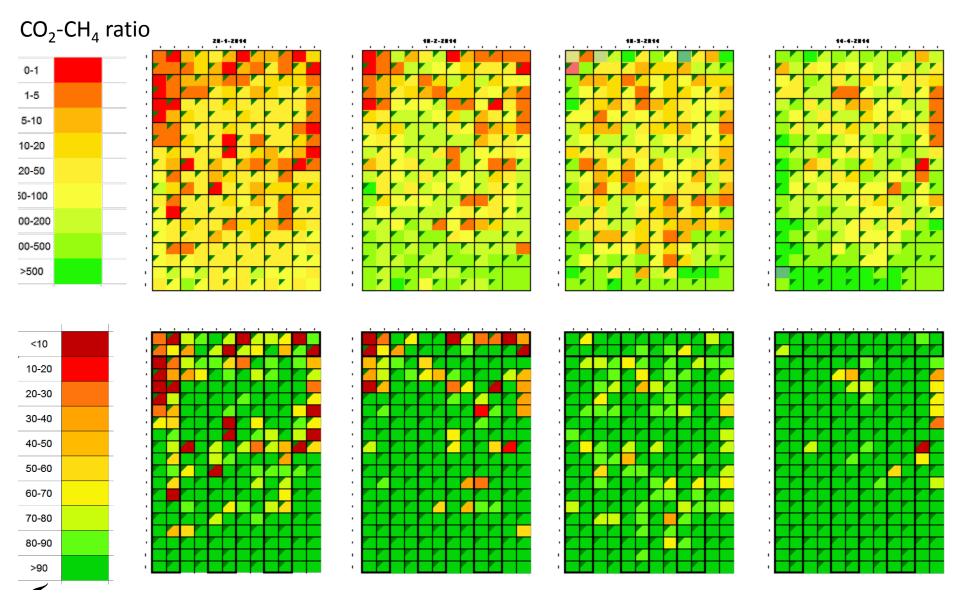




Results of combined CH₄ and CO₂ measurement



Results of combined CH₄ and CO₂ measurement



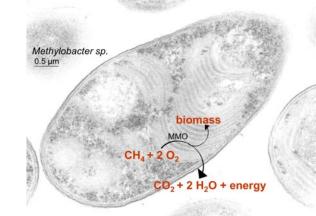
Effciency (% load)

Conclusions factors impacting the process and system performance

Potential oxidation rates of $> 1.200 \,\mathrm{g}$ CH₄ m⁻² d⁻¹ have been reported. Whether this is achieved depends on

- Seasonal changes in temperature and saturation
- Supply of oxygen
- Spatial homogeneity of gas load to system
- Increased load → increased rates (up to a limit)
- Empirical evidence abundant
- MOS can be designed





Project partners in MiMethox

















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